

6-(2-Hydroxybenzoyl)-2-(4-nitrophenyl)pyrazolo-[1,5-a]pyrimidine

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Key indicators

Single-crystal X-ray study
 $T = 150\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.005\text{ \AA}$
 R factor = 0.047
 wR factor = 0.124
Data-to-parameter ratio = 8.3

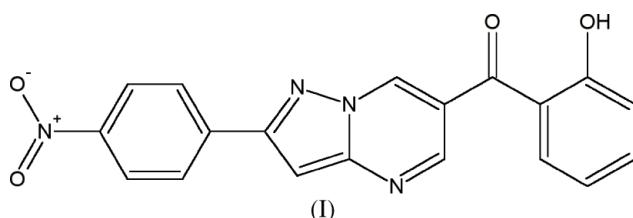
For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The structure of $\text{C}_{19}\text{H}_{12}\text{N}_4\text{O}_4$ contains one strong intramolecular O—H···O bond giving an $S(6)$ motif. The molecules are linked together by weak C—H···O and C—H···N hydrogen bonds, forming a complex three-dimensional network of chain and ring motifs.

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Comment

Since becoming readily available, 3-formylchromone has been used to prepare a variety of heterocyclic systems (Jones & Albrecht, 1976; Haas *et al.*, 1981; Pene & Hubert-Habart, 1980; Sigg *et al.*, 1982). In our investigation of pyrazolo[1,5-a]pyrimidines, we have established that the cyclocondensation reaction of 5-amino-1*H*-pyrazoles with α,β -unsaturated aromatic ketones is a versatile and efficient method for the preparation of these compounds (Orlov *et al.*, 1988; Quiroga *et al.*, 1994, 1999), but the reaction with 3-formylchromone was not previously investigated. We have recently applied the above methodology to prepare several pyrazolo[1,5-a]pyrimidines starting from 3-formylchromone.

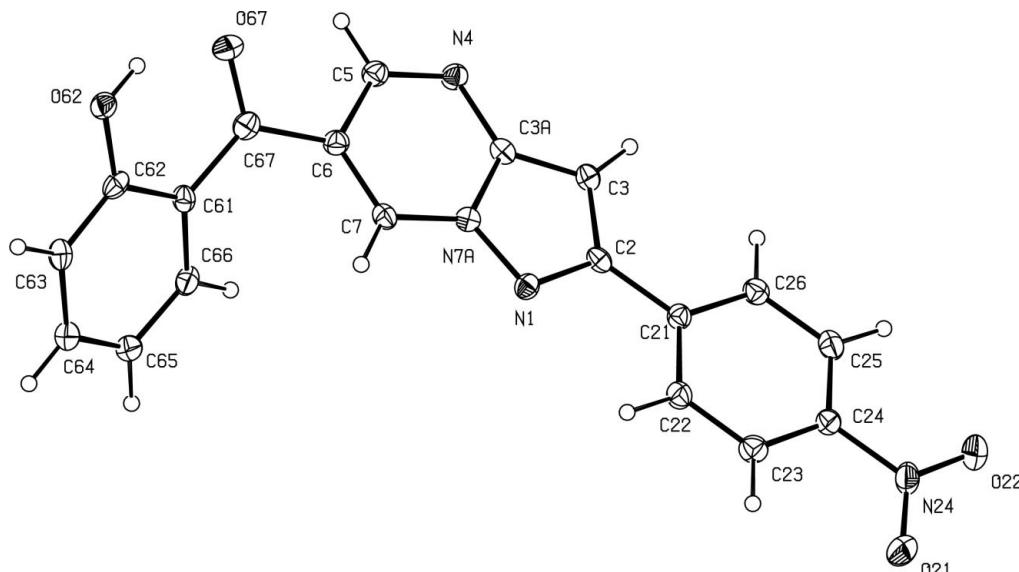


Geometric and hydrogen-bond parameters for (I) are given in Tables 1 and 2, respectively. The three-dimensional hydrogen-bonding network produced by the weak intermolecular C—H···O and C—H···N hydrogen bonds is very complex, the simpler motifs being three infinite chains based on the following motifs; a $C(7)$ motif (Bernstein *et al.*, 1995) involving $\text{C}5-\text{H}5\cdots\text{O}62^i$, a $C(5)$ motif involving $\text{C}66-\text{H}66\cdots\text{O}67^{iii}$ and a $C_2(12)$ motif involving $\text{C}66-\text{H}66\cdots\text{O}67^{iii}$ and $\text{C}63-\text{H}63\cdots\text{N}1^{ii}$ [symmetry codes: (i) $x - 1, y, z$; (iii) $1 - x, y + \frac{1}{2}, -z + \frac{1}{2}$]. The chains produced by repeat of these motifs create ring motifs which in turn produce a three-dimensional network. A view of the molecule is shown in Fig. 1.

Examination of the structure with *PLATON* (Spek, 2000) showed that there were no solvent-accessible voids in the crystal lattice.

Experimental

An equimolar mixture of 3-formylchromone and 5-amino-3-(4-nitrophenyl)pyrazole in ethanol was heated to reflux for 10 min. The title compound precipitated, was separated by filtration and recrys-

**Figure 1**

A view of the title compound with the atomic numbering scheme. Displacement ellipsoids are drawn at the 30% probability level.

tallized from DMF, affording crystals suitable for X-ray diffraction.
M.p. 553–555 K, yield: 90%.

Crystal data

$C_{19}H_{12}N_4O_4$
 $M_r = 360.33$
Orthorhombic, $P2_12_12_1$
 $a = 5.5421 (2) \text{ \AA}$
 $b = 11.5858 (5) \text{ \AA}$
 $c = 23.8057 (13) \text{ \AA}$
 $V = 1528.56 (12) \text{ \AA}^3$
 $Z = 4$
 $D_x = 1.566 \text{ Mg m}^{-3}$

Data collection

KappaCCD diffractometer
 φ and ω scans with κ offsets
Absorption correction: multi-scan (*DENZO-SMN*; Otwinowski & Minor, 1997)
 $T_{\min} = 0.978$, $T_{\max} = 0.992$
7572 measured reflections

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.047$
 $wR(F^2) = 0.124$
 $S = 1.07$
2022 reflections
244 parameters
H-atom parameters constrained

Mo $K\alpha$ radiation
Cell parameters from 7067 reflections
 $\theta = 1.0\text{--}27.5^\circ$
 $\mu = 0.11 \text{ mm}^{-1}$
 $T = 150 (1) \text{ K}$
Needle, yellow
0.20 × 0.08 × 0.07 mm

2022 independent reflections
1364 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.037$
 $\theta_{\text{max}} = 27.4^\circ$
 $h = -7 \rightarrow 7$
 $k = -15 \rightarrow 15$
 $l = -30 \rightarrow 30$

$$w = 1/[\sigma^2(F_o^2) + (0.0480P)^2 + 0.3279P]$$

where $P = (F_o^2 + 2F_c^2)/3$

$$(\Delta/\sigma)_{\text{max}} = 0.005$$

$$\Delta\rho_{\text{max}} = 0.25 \text{ e \AA}^{-3}$$

$$\Delta\rho_{\text{min}} = -0.27 \text{ e \AA}^{-3}$$

Table 2
Hydrogen-bonding geometry (\AA , $^\circ$).

$D-H \cdots A$	$D-H$	$H \cdots A$	$D \cdots A$	$D-H \cdots A$
$C5-H5 \cdots O62^i$	0.95	2.56	3.301 (4)	135
$C63-H63 \cdots N1^{ii}$	0.95	2.58	3.519 (5)	168
$C66-H66 \cdots O67^{iii}$	0.95	2.47	3.332 (4)	150
$O62-H62 \cdots O67$	0.90	1.78	2.583 (4)	147

Symmetry codes: (i) $x - 1, y, z$; (ii) $2 - x, y - \frac{1}{2}, \frac{1}{2} - z$; (iii) $1 - x, \frac{1}{2} + y, \frac{1}{2} - z$.

H atoms were treated as riding with distances $C-H = 0.95 \text{ \AA}$ and $O-H = 0.90 \text{ \AA}$.

Data collection: *KappaCCD Server Software* (Nonius, 1997); cell refinement: *DENZO-SMN* (Otwinowski & Minor, 1997); data reduction: *DENZO-SMN*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEPII* (Johnson, 1976) and *PLATON* (Spek, 2000); software used to prepare material for publication: *SHELXL97* and *WordPerfect* macro *PRPKAPPA* (Ferguson, 1999).

X-ray data were collected at the EPSRC, X-ray Crystallographic Service, University of Southampton, using an Enraf–Nonius KappaCCD diffractometer. The authors thank the staff for all their help and advice. We are grateful to the Ministerio de Educación y Cultura for the award of a grant to one of the authors (AQ).

References

- Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). *Angew. Chem. Int. Ed. Engl.* **34**, 1555–1573.
- Ferguson, G. (1999). *PRPKAPPA*. University of Guelph, Canada.
- Haas, G., Stanton, J. L., Von Sprecher, A. & Wenk, P. (1981). *J. Heterocycl. Chem.* **18**, 607–612.

Table 1
Selected geometric parameters (\AA , $^\circ$).

$N1-C2$	1.341 (5)	$C3A-N4$	1.362 (5)
$N1-N7A$	1.353 (4)	$C3A-N7A$	1.391 (5)
$C24-N24$	1.460 (5)	$C7-N7A$	1.356 (5)
$C2-N1-N7A$	103.6 (3)	$C5-N4-C3A$	115.9 (3)
$O21-N24-O22$	123.3 (3)	$N1-N7A-C7$	125.1 (3)
$O21-N24-C24$	118.7 (3)	$N1-N7A-C3A$	112.6 (3)
$O22-N24-C24$	118.0 (3)	$C7-N7A-C3A$	122.2 (3)

- Johnson, C. K. (1976). *ORTEPII*. Report ORNL-5138. Oak Ridge National Laboratory, Tennessee, USA.
- Jones, W. D. & Albrecht, W. L. (1976). *J. Org. Chem.* **41**, 706–707.
- Nonius (1997). *KappaCCD Server Software*. Windows 3.11 Version. Nonius BV, Delft, The Netherlands.
- Orlov, V. D., Quiroga, J., Kolos, N. N. & Desenko, S. M. (1988). *Khim. Geterosikl. Soedin.* pp. 962–965.
- Otwinowski, Z. & Minor, W. (1997). *Methods Enzymol.* **276**, 307–326.
- Pene, C. & Hubert-Habart, M. (1980). *J. Heterocycl. Chem.* **17**, 329–332.
- Quiroga, J., Insuasty, B., Hormaza, A., Gamenara, D., Domínguez, L. & Saldaña, J. (1999). *J. Heterocycl. Chem.* **36**, 11–13.
- Quiroga, J., Insuasty, B., Rincón, R., Larrahondo, M., Hanold, N. & Meier, H. (1994). *J. Heterocycl. Chem.* **31**, 1333–1335.
- Sheldrick, G. M. (1997). *SHELXS97* and *SHELXL97*. University of Göttingen, Germany.
- Sigg, I., Haas, G. & Winkler, T. (1982). *Helv. Chim. Acta*, **65**, 275–279.
- Spek, A. L. (2000). *PLATON*. May 2000 Version. University of Utrecht, The Netherlands.